We will learn: Metals and Acids

Element			More ambitious			
1.	Metals and Acids	List the metals in order of how well they reacted with acid		ite word equations for the chemical ctions	che	te formula equations for emical reactions if given word uation
2.	Metals and Oxygen	List the metals in order of how well they reacted with oxygen		ite word equations for the chemical ctions	che	te formula equations for emical reactions if given word uation
3.	Metals and Water	List the metals in order of how well they reacted with water		Write word equations for the chemical reactions		scribe the pattern between ctivity and position in the iodic table
					che	te formula equations for mical reactions if given word ation
4.	Metal Displacement	State which metal is more reactive in a pair of named metals		Vrite word equations for the chemical eactions		edict which combinations of tals and metal compounds wil d to displacement reactions.
					che	te formula equations for emical reactions if given word uation
5.	Extracting Metals	Describe how to extract copper from copper oxide.		Explain how copper is extracted from ore.	n its	Explain why carbon is used t extract metals
				Write word equations for the chemica reactions	al	
6.	Ceramics	Describe some uses of ceramics		Construct a list of physical and chem properties of ceramics	ical	Explain how the structure giv ceramics these properties

7	Polymers	Describe the differences between a synthetic and a natural polymer	Explain how polymer properties make them suitable for their uses Select the best polymer for a given job	Explain the properties of different polymers
8	Composites	Describe some properties of composites	Explain how composites properties make them so useful	Plan a fair test into strengths of concrete